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# Synthesis of 7-oxa-phomopsolide E and its C-4 epimer

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Abstract—A flexible, enantioselective route to highly functionalized  $\alpha$ ,  $\beta$ -unsaturated  $\delta$ -lactones has been applied to the synthesis of 7-oxa-phomopsolide E and its C-4 epimer. This approach relies on the application of the Noyori asymmetric hydrogenation of furyl ketone to produce the secondary furyl alcohol in high enantioexcess, which can be stereoselectively transformed into  $\alpha$ , $\beta$ -unsaturated-d-lactones by a short, highly diastereoselective oxidation and reduction sequence. DCC and Mitsunobu coupling were used to introduce the side chains of the two natural product analogs.

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# 1. Introduction

The American elm tree thrives in a wide range of soils and climatic conditions, making it one of the most widely planted street trees in North America.<sup>1</sup> Unfortunately, a completely healthy 100-year-old American elm can be killed in as little as two weeks if infected with Dutch elm disease, $2$  and over 40 million elm trees in the United States have been killed since the arrival of Dutch elm disease.3 The fungus responsible for Dutch elm is spread from tree to tree by the elm bark beetle (Scolytid beetle).

The devastation caused by Dutch elm disease brought about research to find natural compounds that have antiboring/antifeeding activity against the elm bark beetle. As a result,  $G$ rove,<sup>4</sup> and later Stierle,<sup>5</sup> found a series of related 6-substituted 5,6-dihydro-5-hydroxypyran-2-ones, the phomopsolides (Fig. 1), which possess the desired deterrent activity for Scolytid beetles as well as potent antimicrobial activity against Staphylococcus aureus.

As part of a continuing program directed at the discovery of new antibiotics, we have investigated the enantioselective syntheses of various biologically important pyranone natural products both from dienoates<sup>6</sup> and furan alcohols.<sup>7,8</sup> In the case of the phomopsolide class of natural products, our furan



Figure 1. Structures of phomopsolides.

methodology was successfully applied to the synthesis of phomopsolide C  $(1c)$ .<sup>9</sup> Our approach derives its asymmetry from (S)-lactic acid and the application of the Sharpless asymmetric dihydroxylation of vinylfuran. The resulting secondary furan alcohol is produced in high enantioexcess and can be stereoselectively transformed into  $\alpha$ ,  $\beta$ -unsaturated- $\delta$ -lactones via a short highly diastereoselective oxidation and reduction sequence. In our previous synthesis of phomopsolide C (Scheme 1), a Wittig olefination reaction was used to introduce the side chain and was further elaborated into the target structure.

Because of our motivation to test similar compounds as antibiotics and against Dutch elm disease, we felt that a related approach to our synthesis could easily generate structural analogs of the phomopsolide family for biological testing. To these ends the syntheses of the two C-4 diastereomers of 7-oxa-phomopsolide E (7a and 7b) were undertaken. The two targets can easily be envisioned being prepared in a convergent manner from the

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Scheme 1. Retrosynthesis of phomopsolide C.

three intermediates pyranone 3, protected lactic acid 6, and tiglic acid 8 (Scheme 2).



Scheme 2. Retrosynthetic analysis of 7-oxa-phomopsolide D.

In our previous furan approaches we derived the pyranone asymmetry by means of the Sharpless asymmetric dihydroxylation of vinylfuran. $8,10$  Because the enantioselectivity of the dihydroxylation reaction was less than perfect,<sup>11</sup> we searched for a better asymmetric reaction and settled on the Noyori reduction of acylfurans. Thus, the furyl alcohol 5 was enantioselectively prepared from a Noyori asymmetric hydrogenation of furyl ketone 12, which occurred in high yield (84%) and ideal enantioexcess (>96% ee). In turn, the acylfuran 12 could easily be prepared in three steps from glycolic acid 9 via a 2-lithiofuran addition to the TBS-protected amide 11 (Scheme 3). $12$ 

The synthesis of both 7-oxa-phomopsolide E 7a and its 4-epi isomer 7b began with the conversion of furyl



**Scheme 3.** Reagents and conditions: (a) pyrrolidine, xylene,  $150^{\circ}$ C (83%); (b) TBSCl, DMF, Imid., rt (83%); (c) furan/n-BuLi, THF,  $-78 \,^{\circ}\text{C}$  (87%); (d) (1S,2S) Noyori catalyst (13), HCO<sub>2</sub>H/Et<sub>3</sub>N (1:1)  $(84%)$ .

alcohol 5 to pyranone 3 (Scheme 4). Treatment of furyl alcohol  $5$  with  $NBS<sup>13</sup>$  in aqueous THF (Achmatowicz reaction),<sup>14</sup> gave pyranone  $14$  in 92% yield. The reaction of 14 with Jones reagent gave a ketolactone intermediate, which was taken on without purification to a Luche reduction<sup>15</sup> with NaBH<sub>4</sub> and CeCl<sub>3</sub> in MeOH, to give pyranone 3 in 83% yield from 14.



Scheme 4. Reagents and conditions: (a) NBS,  $NaHCO<sub>3</sub>$ ,  $NaOAc$ , THF,  $H_2O$ ,  $0^{\circ}C$  (92%); (b) Jones reagent, acetone,  $0^{\circ}C$ , then NaBH<sub>4</sub>/ CeCl<sub>3</sub>, MeOH/CH<sub>2</sub>Cl<sub>2</sub>,  $-78$  °C (83%).

The desired carboxylic acid  $6$  was synthesized from  $(S)$ ethyl lactate 15 in two steps and 78% yield. This sequence involved hydroxyl group protection of 15 with TBDPSCl to give 16 followed by ester hydrolysis to form the acid 6 (Scheme 5).



Scheme 5. Reagents and conditions: (a) TBDPSCl, DMF, Imid., rt (95%); (b) KOH, MeOH, rt (82%).

The synthesis of 7-oxa-phomopsolide E 7a was then pursued using a DCC-mediated coupling between pyranone 3 and tiglic acid 8 to yield ester 17a (Scheme 6). A subsequent TBS-group deprotection using HF gave a 74% yield (two steps) of alcohol 18a. Once again, resorting to a DCC coupling reaction establishes the remaining carbons of analog 7a. Thus, a DCC coupling of 18a with acid 6 resulted in the protected natural product analog 19a, which was deprotected with HF/Py (2:1) complex providing 7-oxa-phomopsolide E 7a in 91% yield.



Scheme 6. Reagents and conditions: (a)  $8$ , DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub> (85%); (b) HF, CH<sub>3</sub>CN (87%); (c) 6, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub> (89%); (d) HF/Py (2:1), CH<sub>3</sub>CN (91%).



Scheme 7. Reagents and conditions: (a) 8, PPh<sub>3</sub>, DEAD (61%); (b) HF, CH<sub>3</sub>CN (82%); (c) 6, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub> (86%); (d) HF/Py  $(2:1)$ , CH<sub>3</sub>CN (88%).

The 4-*epi* isomer of 7a (7b) was similarly prepared beginning with the esterification of pyranone 3 and tiglic acid 8 but this time under the Mitsunobu conditions (Scheme  $7$ ).<sup>16</sup> Thus simply exposing pyranone 3 and tiglic acid  $\boldsymbol{8}$  to a mixture of PPh<sub>3</sub> and DEAD gave ester 17b in a 61% yield. Ester 17b was deprotected with HF to give a 82% yield of alcohol 18b. DCC coupling between alcohol 18b and acid 6 gave protected 4-epi-7 oxa-phomopsolide E 19b, and HF/Py  $(2:1)$  deprotection gave the natural product analog 7b in 88% yield.

In conclusion, the highly enantio- and diastereocontrolled route to the  $\alpha$ ,  $\beta$ -unsaturated  $\delta$ -lactone natural products developed in our laboratory was successfully applied to the syntheses of the natural product analogs 7-oxa-phomopsolide E and its 4-epi isomer, thus demonstrating the flexibility of the method. The synthesis of both 7a and 7b were completed in only six steps from furan 5 (10 steps from glycolic acid).

#### 2. Experimental section

## 2.1. 1-(2'-Furyl)-2-tert-butyldimethylsilanyloxyethanone (12)

A solution of 2-lithiofuran (0.5 M, 26 ml, 13 mmol) was added dropwise to a solution of amide 11 (2.43 g, 10 mmol) in THF (50 mL) at  $-78$  °C. After stirring for 1 h, the reaction was quenched by addition of satd  $NH<sub>4</sub>Cl$  (20 mL). It was then diluted with Et<sub>2</sub>O (200 mL) and water (100 mL). The organic layer was separated, washed with brine (50 mL), and dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ . Concentration afforded a residue that was purified by flash chromatography (10% EtOAc/hexane) to give the desired ketone 12 (2.09, 8.7 mmol, 87%), as a light yellow oil:  $R_f$  (20% EtOAc/hexane) = 0.61; IR (thin film,  $\text{cm}^{-1}$ ) 2952, 2929, 2856, 1698, 1471, 1255, 1150, 1017, 839; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.58 (dd,  $J = 1.7$ , 0.7 Hz, 1H), 7.32 (dd,  $J = 3.5$ , 0.7 Hz, 1H), 6.54 (dd,  $J = 3.5, 1.7$  Hz, 1H), 4.73 (s, 2H), 0.94 (s, 9H), 0.13 (s, 6H); <sup>13</sup>C NMR (68 MHz, CDCl<sub>3</sub>)  $\delta$  187.0, 150.9, 146.3, 118.0, 112.1, 67.1, 25.8, 18.5,  $-5.4$ ; CI HRMS calcd for  $[C_{12}H_{20}O_3Si+Na]^+$ : 263.1074, found: 263.1076.

# 2.2. 1-(2'-Furyl)-2-tert-butyldimethylsilanyloxyethan-1S-ol (5)

To a 20 ml flask was added ketone 12 (1.42 g, 5.9 mmol), formic acid/triethylamine (1:1, 8 ml),  $CH_2Cl_2$  (3 mL), and Noyori asymmetric transfer hydrogenation catalyst  $(R)$ -Ru( $\eta^6$ -mesitylene)- $(S, S)$ -TsDPEN 13 (18 mg,  $0.5 \,\mathrm{mol}$ %). The resulting solution was stirred at room temperature for 24 h. The mixture was diluted with water  $(20 \text{ mL})$  and extracted with EtOAc  $(3 \times 15 \text{ mL})$ . The organic layers were combined, washed with satd  $NaHCO<sub>3</sub>$ and brine, dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated under reduced pressure to afford the crude alcohol. Flash chromatography (30% EtOAc/hexane) on silica gel yielded  $1.20 \text{ g}$  (5.0 mmol, 84%) of alcohol 5 as a light yellow oil:  $R_f$  (30% EtOAc/hexane) = 0.54;  $[\alpha]_D^{25}$  -15.4° (c 1.01, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm<sup>-1</sup>) 3447, 2954, 2930, 2884, 2857, 1471, 1473, 1361; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$ 7.37 (dd,  $J = 1.7$ , 0.7 Hz, 1H), 6.34 (dd,  $J = 3.2$ , 1.7 Hz, 1H), 6.30 (dd,  $J = 3.2$ , 0.7 Hz, 1H), 4.75 (dd,  $J = 6.4$ , 4.5 Hz, 1H), 3.85 (dd,  $J = 10.1$ , 4.5 Hz, 1H), 3.83 (dd,  $J = 10.1$ , 6.4 Hz, 1H), 2.89 (d,  $J = 4.2$  Hz, 1H), 0.89 (s, 9H), 0.06 (s, 6H); <sup>13</sup>C NMR (68 MHz, CDCl<sub>3</sub>)  $\delta$  153.6, 142.0, 110.2, 107.0, 68.3, 65.6, 25.8, 18.2,  $-5.5$ ; calcd for  $[C_1,H_2,O_3Si-H_2O]^+$ : 225.1310, found: 225.1296.

# 2.3. (5S,6S)-5-(2-Methyl-2-butenoyloxy)-6-(tert-butyldimethylsilanyloxymethyl)-5,6-dihydropyran-2-one (17a)

Alcohol 3 (136 mg, 0.53 mmol), tiglic acid (105 mg, 1.05 mmol), dicyclohexylcarbodiimide (217 g, 1.05 mmol), and dimethylaminopyridine (catalytic amount) were dissolved in 64 ml of  $CH_2Cl_2$ . The reaction mixture was stirred at room temperature for 5 h. The reaction mixture was then filtered through a pad of Celite with excess  $Et<sub>2</sub>O$  and concentrated. The crude product was purified by silica gel flash chromatography eluting with 10% EtOAc/hexane to yield 152 mg (0.45 mmol, 85%) of 17a as a clear oil:  $R_f$  (30% EtOAc/hexane) = 0.78;  $[\alpha]_D^{25}$ +250 $\textdegree$  (c 1.10, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm<sup>-1</sup>) 2955, 2930, 2857, 1715, 1253, 1134, 1096, 1068, 837; 1H NMR  $(500 \text{ MHz}, \text{ CDCl}_3)$   $\delta$  7.07 (dd,  $J = 10.0, 6.0 \text{ Hz}, 1\text{ H}$ ), 6.86 (dd,  $J = 14.0$ , 6.0 Hz, 1H), 6.17 (d,  $J = 9.5$  Hz, 1H), 5.33 (dd,  $J = 6.0$  Hz, 2.5 Hz, 1H), 4.56 (ddd,  $J = 7.0$ , 7.0, 2.5 Hz, 1H), 3.88 (d,  $J = 7.0$  Hz, 1H), 1.81–1.76 (m, 6H), 0.83 (s, 9H), 0.01 (d,  $J = 16.5$  Hz, 6H); <sup>13</sup>C NMR  $(125 \text{ MHz}, \text{CDC1}_3)$   $\delta$  166.0, 162.4, 141.0, 139.3, 127.7, 124.9, 78.5, 60.9, 60.2, 25.7, 18.1, 14.5, 12.0,  $-5.5$ ,  $-5.6$ ; CI HRMS calcd for  $[C_{17}H_{28}O_5Si+H]^+$ : 341.1784, found: 341.1772.

## 2.4. (5S,6S)-5-(2-Methyl-2-butenoyloxy)-6-hydroxymethyl-5,6-dihydropyran-2-one (18a)

Ester  $17a$  (140 mg, 0.41 mmol), 1 mL of CH<sub>3</sub>CN, and HF  $(5\%$ , 2 ml,  $\sim$ 5.0 mmol) were added to a plastic vial and stirred at rt for 10 h. The reaction was quenched with saturated  $NaHCO<sub>3</sub>$ , the aqueous layer was extracted with EtOAc  $(2 \times 30 \text{ mL})$ , and the organic layer was dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated under reduced pressure to afford the crude alcohol. Flash chromatography (60%

EtOAc/hexane) on silica gel yielded 81 mg (0.36 mmol, 87%) of alcohol 18a as a colorless oil:  $R_f$  (50% EtOAc/ hexane) = 0.23;  $[\alpha]_D^{25}$  +285° (c 1.02, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm1) 3446, 2934, 1712, 1649, 1382, 1256, 1131, 1066; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.02 (dd,  $J = 9.6$ , 6.0 Hz, 1H), 6.95–6.85 (m, 1H), 6.24 (d,  $J = 9.6$  Hz, 1H), 5.41 (dd,  $J = 6.0$ , 2.7 Hz, 1H), 4.64 (ddd,  $J = 6.5, 6.5$ , 2.7 Hz, 1H), 3.94 (dd,  $J = 12.0$  Hz, 6.9 Hz, 1H), 3.75 (dd,  $J = 12.0$  Hz, 6.3 Hz, 1H), 1.82–1.78 (m, 6H); <sup>13</sup>C NMR  $(75 \text{ MHz}, \text{ CDC1}_3)$   $\delta$  167.1, 162.3, 140.2, 139.1, 127.1, 125.0, 79.1, 61.9, 60.3, 14.5, 11.9; ESI HRMS calcd for  $[C_{11}H_{14}O_5 + Na]^+$ : 249.0739, found: 249.0751.

# 2.5. (5S,6S)-5-(2-Methyl-2-butenoyloxy)-6-[(2S)-2-(tertbutyldiphenylsilanyloxy)propionyloxymethyl]-5,6-dihydro-pyran-2-one (19a)

Alcohol 18a (79 mg, 0.35 mmol), (2S)-(tert-butyldiphenylsilanoxy)propionic acid (228 mg, 0.70 mmol), dicyclohexylcarbodiimide (143 mg, 0.70 mmol), and dimethylaminopyridine (catalytic amount) were dissolved in  $8 \text{ ml}$  of  $CH_2Cl_2$ . The reaction mixture was stirred at room temperature for 6 h. The reaction mixture was then filtered through a pad of Celite with  $Et<sub>2</sub>O$ (40 mL) and concentrated. The crude product was purified by silica gel flash chromatography eluting with 20% EtOAc/hexane to yield 164 mg (0.31 mmol, 89%) of 19a as a colorless oil:  $R_f$  (30% EtOAc/hexane) = 0.47;  $[\alpha]_{\text{D}}^{25}$  +107° (c 1.02, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm<sup>-1</sup>) 3069, 2962, 2936, 2860, 1750, 1717, 1651, 1428, 1248, 1135, 824; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  7.69–7.62 (m, 4H), 7.44 (m, 6H), 6.99 (dd,  $J = 9.7$ , 5.7 Hz, 1H), 6.87 (m, 1H), 6.20 (d,  $J = 9.7$  Hz, 1H), 5.03 (dd,  $J = 5.7$ , 2.7 Hz, 1H), 4.46 (ddd,  $J = 6.4$ , 6.4, 2.7 Hz, 1H), 4.31 (q,  $J = 6.7$  Hz, 1H), 4.20 (m, 2H), 1.83–1.79 (m, 6H), 1.38 (d,  $J = 6.7$  Hz, 3H), 1.08 (s, 9H); <sup>13</sup>C NMR (68 MHz, CDCl<sub>3</sub>)  $\delta$  173.2, 166.3, 161.6, 140.3, 140.0, 135.9, 135.7, 133.2, 133.0, 129.9, 127.7, 127.6, 127.2, 124.8, 75.8, 68.6, 61.5, 61.0, 26.7, 21.2, 19.2, 14.6, 12.0; calcd for  $[C_{30}H_{36}O_7Si+Na]^+$ : 559.2123, found: 559.2137.

#### 2.6. 7-Oxa-phomopsolide E (7a)

Ester 19 (153 mg, 0.29 mmol), 2 ml of CH<sub>3</sub>CN, and HF/ Py (2:1) (2.5 M, 3 mL,  $\sim$  7.5 mmol) were added to a plastic vial and stirred at rt for 24 h. The reaction was quenched with saturated  $NAHCO<sub>3</sub>$ , and the aqueous layer was extracted with EtOAc  $(2 \times 40 \text{ mL})$ . The organic layer was washed with HCl (1 M, 10 mL), dried over  $Na<sub>2</sub>SO<sub>4</sub>$ , and concentrated under reduced pressure to afford the crude alcohol. Flash chromatography (60% EtOAc/hexane) on silica gel yielded 78 mg (0.26 mmol, 91%) of 7-oxa-phomopsolide E 7a as a colorless oil:  $R_f$  (50% EtOAc/hexane) = 0.21;  $[\alpha]_D^{25}$  +243° (c 1.20, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm<sup>-1</sup>) 3468, 2933, 1715, 1648, 1450, 1381, 1254, 1132, 1099, 1068; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>)  $\delta$  7.05 (dd,  $J = 9.5, 5.9$  Hz, 1H), 6.99–6.87 (m, 1H), 6.27 (d,  $J = 9.9$  Hz, 1H), 5.40 (dd,  $J = 5.9$ , 3.0 Hz, 1H), 4.85  $(\text{ddd}, J = 6.0, 3.0, 0.9 \text{ Hz}, 1\text{H}), 4.47 \text{ (ddd}, J = 11.7, 6.0,$ 6.0, 2H), 4.34 (q,  $J = 6.9$  Hz, 1H), 1.87–1.79 (m, 6H), 1.43 (d,  $J = 6.9$  Hz, 3H); <sup>13</sup>C NMR (75 MHz, CDCl<sub>3</sub>) d 174.7, 166.2, 161.5, 140.4, 138.9, 127.0, 124.8, 76.6, 66.6, 62.3, 61.2, 20.0, 14.3, 11.9; ESI HRMS calcd for  $[C_{14}H_{18}O_7 + Na]^+$ : 321.0950, found: 321.1188.

## 2.7. (5R,6S)-5-(2-Methyl-2-butenoyloxy)-6-(tert-butyldimethylsilanyloxymethyl)-5,6-dihydropyran-2-one (17b)

Alcohol 3 (235 mg, 0.91 mmol) was dissolved in 6 mL of benzene. The solution was cooled to  $0^{\circ}$ C and triphenylphosphine (358 mg, 1.37 mmol), tiglic acid (136 mg, 1.36 mmol), and diethyl azodicarboxylate (238 mg, 1.37 mmol) were added to the solution. The solution was stirred for 12 h, quenched with saturated aqueous sodium bicarbonate (30 mL), and extracted with EtOAc  $(2 \times 30 \text{ mL})$ . The organic fractions were combined, washed with brine (30 mL), dried  $(Na_2SO_4)$ , and concentrated. Purification on silica gel (EtOAc/hexane, 3:7) yielded 188 mg  $(0.55 \text{ mmol}, 61\%)$  of ester 17b as a colorless oil:  $R_f$  (30% EtOAc/hexane) = 0.65;  $[\alpha]_D^{25}$  -155° (c 1.20, CH<sub>2</sub>Cl<sub>2</sub>); IR (thin film, cm<sup>-1</sup>) 2954, 2932, 2856, 1742, 1716, 1651, 1253, 1130, 837; <sup>1</sup>H NMR (270 MHz, CDCl<sub>3</sub>)  $\delta$  6.93 (q,  $J = 6.9$  Hz, 1H), 6.82 (dd,  $J = 9.9$ , 3.7 Hz, 1H), 6.10 (dd,  $J = 9.9$ , 1.2 Hz, 1H), 5.60 (ddd,  $J = 5.2, 4.0, 1.2$  Hz, 1H), 4.57 (dd,  $J = 7.9, 4.0$  Hz, 1H), 3.85 (dd,  $J = 4.2$ , 1.5 Hz, 2H), 1.83–1.80 (m, 6H), 0.85  $(s, 9H)$ , 0.04  $(s, 6H)$ ; <sup>13</sup>C NMR (68 MHz, CDCl<sub>3</sub>) d 166.7, 161.9, 142.1, 139.4, 127.6, 122.9, 80.4, 63.4, 62.5, 25.7, 18.2, 14.5, 12.0,  $-5.6$ ; calcd for  $[C_{30}H_{36}O_7Si+Na]^+$ : 559.2123, found: 559.2133.

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### References and notes

- 1. Becker, H. Agric. Res. 1996, 44, 4–8.
- 2. Santamour, F. S.; Bentz, S. E. J. Arboric. 1995, 21, 122–131.
- 3. Sinclair, W. A.; Lyon, H. H.; Johnson, W. T. In Diseases of Trees and Shrubs; Cornell University Press: Ithaca, NY, 1987; p 574.
- 4. Grove, J. F. J. Chem. Soc., Perkin Trans. 1 1985, 865–869.
- 5. Stierle, D. B.; Stierle, A. A.; Ganser, B. J. Nat. Prod. 1997, 60, 1207–1209.
- 6. For the synthesis of cryptocarya diacetate and cryptocarya triacetate, see: (a) Hunter, T. J.; O'Doherty, G. A. Org. Lett. 2001, 3, 2777-2780; (b) Smith, C. M.; O'Doherty, G. A. Org. Lett. 2003, 5, 1959–1962.
- 7. Harris, J. M.; O'Doherty, G. A. Tetrahedron Lett. 2000, 41, 183–187.
- 8. For the application of this approach toward several styryllactone natural products, see: (a) Harris, J. M.; O'Doherty, G. A. Tetrahedron 2001, 57, 5161-5171; (b) Harris, J. M.; O'Doherty, G. A. Org. Lett. 2000, 2, 2983– 2986.
- 9. Harris, J. M.; O'Doherty, G. A. Tetrahedron Lett. 2002, 43, 8195–8199.
- 10. Harris, J. M.; O'Doherty, G. A. Tetrahedron Lett. 2000, 41, 183–187.
- 11. The Sharpless reaction of vinylfuran occurs with enantioexcess on the order of 90% and requires a re-crystallization of the corresponding bis-benzoates to get enantiomerically pure diol, see: Harris, J. M.; Keranen, M. D.; Nguyen, H.; Young, V. G.; O'Doherty, G. A. Carbohydr. Res. 2000, 328, 17–36.
- 12. For a related procedure using lactic acid, see: (a) Ferrero, M.; Galobardes, M.; Martin, R.; Montes, T.; Romea, R.; Rovira, R.; Urpi, F.; Vilarrasa, J. Synthesis 2000, 1608– 1614; (b) Fein, M. L.; Filachione, E. M. J. Am. Chem. Soc. 1953, 75, 2097–2099.
- 13. (a) Grapsas, I. K.; Couladouros, E. A.; Georgiadis, M. P. Polym. J. Chem. 1990, 64, 823–826; (b) Georgiadis, M. P.; Couladoros, E. A. J. Org. Chem. 1986, 51, 2725–2727.
- 14. An Achmatowicz reaction is the oxidative rearrangement of furfuryl alcohols to 2-substituted 6-hydroxy-2H-pyran-3(6H)-ones, see: (a) Ref. 13, Achmatowicz, O.; Bielski, R. Carbohydr. Res. 1977, 55, 165–176; (b) For its use in carbohydrate synthesis, see Ref. 12 and Balachari, D.; ODoherty, G. A. Org. Lett. 2000, 2, 863–866; (c) Balachari, D.; O'Doherty, G. A. Org. Lett. 2000, 2, 4033–4036.
- 15. Luche, J.-L. J. Am. Chem. Soc. 1978, 110, 2226–2227.
- 16. Previously we have shown that the Mitsunobu esterification reaction works well on pyranone 3, see Ref. 7, Mitsunobu, O. Synthesis 1981, 1.